IUPAC Task Group on Atmospheric Chemical Kinetic Data Evaluation – Data Sheet oIOx3

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$CH_2IO_2 + CH_2IO_2 \rightarrow CH_2IOH + HCOI + O_2$	(1)
\rightarrow 2CH ₂ IO + O ₂	(2)

Rate coefficient data ($k = k_1 + k_2$)

k/cm^3 molecule ⁻¹ s ⁻¹	Temp./K	Reference	Technique/ Comments
<i>Absolute Rate Coefficients</i> ~9 x 10 ⁻¹¹	295	Sehested, Ellermann, and Nielsen, 1994 ¹	(a)

Comments

(a) Pulse radiolysis UV absorption spectrometric study of CH₃I-O₂-SF₆ mixtures at a total pressure of 1000 mbar. CH₂IO₂ radicals were monitored by UV absorption with $\sigma_{370} = (2.1 \pm 0.5) \times 10^{-18}$ cm² molecule⁻¹. The interpretation of the kinetic data is complicated by the presence of CH₃O₂ radicals, which leads to mixed-order kinetics. The above approximate value of *k* was derived on the basis of several assumptions.

Preferred Values

No recommendation.

Comments on Preferred Values

The approximate and exceptionally large rate coefficient obtained by Sehested *et al.*¹ should be regarded with caution owing to the inherent complications in their experimental system. Until more work is carried out in this reaction, we make no recommendation.

References

¹ J. Sehested, T. Ellermann and O. J. Nielsen, Int. J. Chem. Kinet. 26, 259 (1994).